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(64) **Method of Making Microcapsules**

(57) In the method of making microcapsules, polycondensation for forming an anion-modified aminoaldehyde resin is performed in the presence of at least one anionic

colloid material in an aqueous dispersion including hydrophobic material particles dispersed therein to deposit said anion-modified aminoaldehyde resin on said hydrophobic material particles to form microcapsules encapsulating hydrophobic capsule core material therein.

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SPECIFICATION**Method of Making Microcapsules****Background of the Invention**

This invention relates to a method of making microcapsules containing hydrophobic capsule core material and more particularly to a method of making microcapsules having an improved capsule core material retainability. 5

Microcapsules find their usefulness in various fields such as pressure-sensitive copying papers, heat-sensitive recording papers, adhesives, fertilizers, pharmaceuticals, foods, cosmetics, etc. There are known various methods for making microcapsules useful for those purposes. Among them, there are 10 included the "coacervation" method, the "interfacial polycondensation" method and the "in situ polymerization" method. It is also known that microcapsules each having a capsule wall of an aminoaldehyde resin are superior in the water resistance and the solvent resistance. For example, US Patent Specification No. 3,016,308 discloses a method of making microcapsules in which urea-formaldehyde polycondensation resin capsule walls are formed in the presence of carboxymethyl 15 cellulose. US Patent Specification No. 3,516,441 discloses another method of making microcapsules in which urea-formaldehyde polycondensation resin capsule walls are formed in a suspension substantially containing no wetting agent. In these conventional methods of making microcapsules utilizing urea-formaldehyde resin, however, the system for making microcapsules must be carefully controlled as by stepwise and continual dilution since effective deposition of the polycondensation 20 product on the capsule core material cannot be expected. In order to effectively carry out deposition of the polymerized resin on the capsule core material, some attempts have been made. For example, Japanese Patent Publication No. 12,380 of 1962 proposes to utilize as a dispersing agent a substance having an active radical which is capable of causing chemical or physicochemical bond and each of Japanese Patent Publications Nos. 12,518 of 1963, 4,717 of 1973 and 13,456 of 1974 discloses to 25 utilize phase separation by an electrostatic interaction. In these improved methods, however, the steps for making microcapsules are complicated as well as in the so-called "complex coacervation" method. Those methods also involve a disadvantage that the capsule walls are often cracked in the step of drying. This may be considered due to the fact that the microcapsule walls contain water soluble components having opposite net electrical charges.

30 An object of the invention is to provide a method for making microcapsules each having a capsule wall of an aminoaldehyde resin in which deposition of an aminoaldehyde polycondensation resin on the surface of capsule core material is effectively carried out.

Another object of the invention is to provide a method for performing polycondensation of an anion-modified aminoaldehyde resin in a microcapsule making system to deposit the anion-modified 35 aminoaldehyde resin on individual capsule core material particles for encapsulation.

The invention enables microcapsules having good water resistance and good solvent resistance to be made in a simple and effective manner.

Summary of the Invention

According to the invention, the method for making microcapsules comprises the steps of 40 preparing an aqueous dispersion including hydrophobic material particles dispersed therein, establishing in said dispersion a system for causing polycondensation to form an anion-modified aminoaldehyde resin and performing said polycondensation to deposit said anion-modified aminoaldehyde resin on said hydrophobic material particles to form microcapsules encapsulating said hydrophobic material therein, and characterized in that said polycondensation is carried out in the 45 presence of at least one anionic colloid material. Preferably, the anion-modified aminoaldehyde resin is one modified by an anion-modifying agent selected from the group consisting of sulfamic acid, sulfanilic acid, glycine and combinations of the foregoing.

In a preferred embodiment of the invention, the system for causing polycondensation for forming an anion-modified aminoaldehyde resin comprises a prepolymer of at least one amine and at least one 50 aldehyde.

Detailed Description of the Invention

According to the invention, the microcapsules have a capsule wall of an anion-modified aminoaldehyde resin obtained by polycondensation. Such the microcapsules can be produced by performing the polycondensation to form an anion-modified aminoaldehyde resin in the presence of at 55 least one anionic colloid material in an aqueous dispersion including hydrophobic capsule core material particles dispersed therein. The hydrophobic capsule core material may be either of the liquid phase or of the solid phase. Among typically useful capsule core materials, there may be included: water-insoluble or substantially water-insoluble liquids such as animal oils, e.g. fish oil, and lard oil, vegetable oils, e.g. olive oil, peanut oil, linseed oil, soy-bean oil, and castor oil, mineral oils, e.g., petroleum, kerosene, xylene and toluene, and synthetic oils, e.g. alkylated diphenylethane, alkylated naphthalene, 60 diphenylethane and methyl salicylate; water-insoluble metal oxides and salts; fibrous materials, e.g., cellulose and asbestos; water insoluble synthetic polymers; minerals; pigments; glasses; perfumes; flavourings; germicidal compositions; physiological compounds and fertilizers.

For pressure sensitive copying paper the above-mentioned oils as capsule core materials may include at least one chromogenic material dissolved therein.

According to the invention, an anion-modified aminoaldehyde resin is used as capsule wall forming material. The term "anion-modified aminoaldehyde resin" described means an aminoaldehyde resin modified by a modifier so as to become anionic.

The aminoaldehyde resin is obtained by polycondensation of at least one amine, such as urea, thiourea, alkyl urea, ethylene urea, acetoguanamine, enzoguanamine, melamine, guanidine, dicyandiamine, biuret and cyanamide and at least one aldehyde such as formaldehyde, acetaldehyde, paraformaldehyde, hexamethylenetetramine, butyraldehyde, crotonaldehyde, benzaldehyde and furfural.

The aminoaldehyde resin is modified with an anion-modifier, such as sulfamic acid, sulfanilic acid, glycolic acid, glycine, acidsulfite, phenolsulfonic acid and taurine.

Among the anion-modified aminoaldehyde resin, those in which urea and/or melamine and formaldehyde are used as starting materials and at least one of sulfamic acid, sulfanilic acid and glycine is used as the anion-modifier are preferred because the capsule wall obtained has a greatly improved core material retainability and a greatly improved impermeability.

The above-mentioned anion-modifier may be used in an amount of 1 to 50 mol%, preferably 2 to 20 mol%, with respect to the amount of the amine used.

The anion-modified aminoaldehyde resin used according to the invention may also include any of other polycondensation components such as phenol, m-methoxyphenol, resorcin, pyrogallol and the like for changing the other properties of the capsule wall as desired. As a matter of course, the amount of such other polycondensation components included should be within such the range that the desirable effects according to the invention can be achieved and maintained.

The system for causing polycondensation to form an anion-modified aminoaldehyde resin according to the invention is established in the before-mentioned aqueous dispersion including hydrophobic capsule core material particles dispersed therein. Preferably, the system for causing polycondensation comprises a water soluble prepolymer of at least one amine and at least one aldehyde. In case of using a water soluble prepolymer of at least one amine and at least one aldehyde, the anion-modifier may either previously bonded to the prepolymer or separately added to the aqueous dispersion. However, the system for causing polycondensation may also be established in the aqueous dispersion of capsule core material by adding at least one amine monomer, at least one aldehyde monomer and at least one anion-modifier separately and directly to said aqueous dispersion.

According to the invention, the polycondensation to form an anion-modified aminoaldehyde resin is performed in presence of at least one anionic colloid material in said aqueous dispersion of hydrophobic capsule core material.

Among the typically useful anionic colloid materials, there may be included natural polymers such as gum arabic, carrageenan, sodium alginate, pectic acid, tragacanth gum, almond gum and agar, semisynthetic polymers such as carboxymethylcellulose, sulfonated cellulose, sulfonated methylcellulose, carboxymethylated starch and phosphorated starch and synthetic polymers such as polymers having at least one anionic monomer unit such as acrylic acid, methacrylic acid, maleic acid, crotonic acid, itaconic acid and vinyl-benzene sulfonic acid, partial amidation compound or partial esterification compound thereof and carboxylated polyvinyl alcohol.

Especially in case where the capsule core material is a hydrophobic liquid such as oily material, anionic synthetic polymers having a hydrophobic monomer unit are preferably used for obtaining a stable emulsion. Most preferably, copolymers of hydrophobic monomers such as ethylene, propylene, isobutylene, styrene, methylvinyl ether, vinylacetate, acrylic acid ester, methacrylic acid ester and itaconic acid ester with at least one of acrylic acid, methacrylic acid and maleic acid are advantageously used because the capsule walls of the microcapsules obtained are not substantially deformed and have an improved physical and chemical strength.

The anionic colloid material may be used in an amount of at least 0.1% by weight of the amount of the aqueous dispersion. The preferred amount of the anionic colloid material for obtaining improved microcapsules in a simpler manner would be at least 0.5% by weight, most preferably, at least 2% by weight of the amount of the aqueous dispersion. The upper limit of the amount of the anionic colloid material depends on the device used for making microcapsules but generally may be at largest 20% by weight of the aqueous dispersion.

The polycondensation for forming an anion modified aminoaldehyde resin may preferably be performed in the above mentioned aqueous dispersion in an acidic state. In order to adjust the system for the production of microcapsules in an acidic state any of known acid catalysts such as formic acid, acetic acid, citric acid, oxalic acid, paratoluenesulphonic acid, hydrochloric acid, sulfuric acid, ammonium chloride and ammonium sulfate may be used. For this purpose it is also possible to utilize the acidic radical included in the anionic colloid material which co-exists in the system according to the invention.

In order to accelerate the polycondensation the system for the production of microcapsules may be heated at a temperature between the range of 20 to 100°C, most preferably, within the range of 35

to 70°C. In this manner microcapsules having stable properties can be produced for relatively short time.

According to the invention the polymerized resin is effectively deposited on each of the capsule core material particles. This can be achieved by merely admixing various materials for forming

- 5 microcapsules and providing proper conditions for polycondensation. Careful control of the system as required in the conventional method, for example, as by steadily adding water for dilution, is not needed. The theory and mechanism why the good result is obtained according to the invention are not clear. However, it may be assumed that the polycondensation product can effectively be deposited on the surface of capsule core material with the dispersion system being maintained in the stable
10 condition, due to the repulsive force by electron charges of the anionic colloid material and the produced anionic aminoaldehyde resin.

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Detailed Description of the Preferred Embodiment

The following examples serve to illustrate the invention in more detail although the invention is not limited to the examples. Unless otherwise indicated, parts and % signify parts by weight and % by
15 weight, respectively.

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Example 1

- 2.8 parts of crystal violet lactone and 0.1 parts of benzoyl leucomethylene blue were dissolved in 100 parts of alkyl naphthalene (KMC oil manufactured by Kureha Kagaku Kogyo Kabushiki Kaisha, Japan) to prepare an oily solution. The oily solution was dispersed in 150 parts of a 3% aqueous
20 solution (pH 4.0) of a hydrolyzed ethylene-maleic anhydride copolymer to obtain an emulsion containing oily capsule core material particles which have an average particle size of 3.4 microns. On the other hand 10 parts of melamine was added to 30 parts of a 37% aqueous solution of
25 formaldehyde pH of which was adjusted to 7.0 with sodium hydroxide and the mixture was heated at 70°C for 30 minutes. Then to the mixture 10 parts of water and 1 part of sulfamic acid were added and the resultant aqueous system was heated at 60°C for 2 minutes to prepare an aqueous solution of
30 anion-modified melamine formaldehyde resin prepolymer. The prepolymer solution was slowly added and admixed into the above emulsion maintained at 55°C. Upon addition of the prepolymer solution formation of capsule walls was initiated. The emulsion system was gently agitated at the same temperature for 3 hours to form a microcapsule dispersion in which microcapsules, each having fully
35 cured capsule wall and an average particle size of 3.6 microns, were dispersed.

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Thus obtained microcapsule dispersion was cooled on standing, and then 25 parts of cellulose powder and 200 parts of 2% aqueous solution of hydroxyethylcellulose were added to the microcapsule dispersion to prepare a capsule coating composition. The coating composition was coated on one surface of a paper sheet of 40 g/m² in an amount of 4 g/m² on dry basis and dried to obtain a capsule coated sheet.

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Control 1

- Example 1 was repeated except that polyvinyl alcohol which was nonionic and water soluble colloid material, was used instead of a hydrolyzed ethylene-maleic anhydride copolymer. Resultantly, the emulsion system became very viscous after adding the prepolymer solution and then free oil was found in the system after completing the reaction. A capsule dispersion as desired was not obtained.

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Control 2

- Example 1 was repeated except that an aqueous solution of a cationic gelatin having an isoelectric point of 8.0 was used, under the same capsule preparing condition, instead of the hydrolyzed ethylene-maleic anhydride copolymer. As the prepolymer solution was added, a remarkable tendency
45 to cause agglomeration was observed. The resultant capsule dispersion was used to prepare a capsule 45 coated paper sheet in the same manner as in Example 1.

- Examples 2—5**
4 capsule dispersions were prepared in the same manner as in Example 1 except that anionic colloid materials and anion modifying agents as shown in Table 1 were used. Capsule coated paper sheets were prepared with the use of the capsule dispersions in the same manner as in Example 1.

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Table 1

	<i>Anionic colloid materials</i>	<i>Anion modifying agents</i>	
Example 2	Hydrolyzed product of Isobutylene-maleic anhydride copolymer	sulfanilic acid	
Example 3	Hydrolyzed product of styrene-maleic anhydride copolymer	sulfamic acid	55

Table 1 (contd.)

	<i>Anionic colloid materials</i>	<i>Anion modifying agents</i>	
Example 4	Maleic acid-modified polyvinyl alcohol	glycine	
5 Example 5	Carrageenan	sulfamic acid	5
Example 6			
10	0.5 parts of m-methoxyphenol was added to 150 parts of 5% aqueous solution (pH 3.0) of acrylic acid-methyl acrylate copolymer to prepare an aqueous system. 100 parts of the same oily solution as in Example 1 was dispersed in the aqueous system to prepare an emulsion containing oily capsule core material particles which has an average particle size of 4.0 microns. On the other hand, 5 parts of monochloroacetic acid was dissolved in 80 parts of 37% aqueous solution of formaldehyde and then 7 parts of 25% ammonia water and 4 parts of sodium hydroxide were added to the solution. After the resultant mixture was thoroughly agitated, 30 parts of urea was added to it and the reaction mixture was refluxed for 30 minutes. The pH of the product was adjusted to 7.0 to prepare an aqueous solution of an anion-modified urea-formaldehyde resin prepolymer.	10	
15	21 parts of the thus obtained prepolymer solution was added to the above emulsion. The resultant aqueous system has a pH of 3.4. The aqueous system was heated at 60°C for 3 hours for reaction and then cooled with standing to prepare a capsule dispersion. The capsule dispersion was used to prepare a capsule coated paper sheet in the same manner as in Example 1.	15	

20 Examples 7—9	3 capsule dispersions were prepared in the same manner as in Example 6 except that anionic colloid materials as shown in Table 2 were used and the pH of each of the dispersions after adding the prepolymer was adjusted to 3.4. The capsule dispersions were used to prepare capsule coated sheets in the same manner as in Example 1.	20
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25	Table 2 Anionic Colloid Materials	25
	Example 7 Half-amidation compound of isobutylene-maleic anhydride copolymer	
	Example 8 Carboxymethylcellulose	
	Example 9 Pectic acid	

30 Example 10	5 parts of urea and 0.5 parts of glycine were added to and dissolved in 150 parts of 5% aqueous solution (pH 3.4) of a half-amidation compound of isobutylene-maleic anhydride copolymer to obtain a homogeneous aqueous solution, 100 parts of the same oily solution as in Example 1 was dispersed in the aqueous solution to prepare an emulsion containing oily capsule core material particles which had an average particle size of 4.0 microns. 13 parts of 37% aqueous solution of formaldehyde was added to the emulsion. The mixture was heated to 55°C under a gentle agitation, 5 parts of 10% aqueous solution of resorcin was added to the mixture after maintaining at 55°C for 1 hour and then the mixture was further heated at 55°C for 2 hours and cooled on standing to obtain a capsule dispersion. The resultant capsule dispersion was used to obtain a capsule coated paper sheet.	30
40	The properties of the thus obtained capsule coated paper sheets were examined with the following manners.	40

45	At first, an acceptor coated sheet was prepared by the steps of adding 65 parts of aluminum hydroxide, 20 parts of zinc oxide, 15 parts of a co-melt of zinc, 3,5-di(α -methylbenzyl)salicylate and α -methylstyrene-ethylene copolymer in which the co-melting ratio was 80/20 and an aqueous solution containing 5 parts of polyvinyl alcohol to 300 parts of water, finely dividing the mixture for 24 hours in a ball-mill to obtain a dispersion, adding to the dispersion a latex of carboxylated styrene-butadiene copolymer in which the solid amount was 20 parts to prepare a coating composition, applying the coating composition on one surface of a base paper sheet of 40 g/m ² in an amount of 5 g/m ² on dry basis and drying the coated paper sheet.	45
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50 1. Color formability:	Each capsule coated paper sheet obtained in Examples and Controls was put on the above acceptor coated sheet so that the coated layers were close to each other to provide samples. The samples were pressed with 100 Kg/cm ² for 1 minute to form a color image. The color density of the image on the acceptor coated surface was measured by a spectrophotometer at 600 nm with the use of magnesium oxide as a standard. The measured reflection ratios are shown in Table 3.	50
55 2. Stability of color forming:	Each capsule coated sheet obtained in Examples and Controls was treated by standing under the following conditions:	55

60	(a) heat resistance (b) solvent resistance (c) humidity resistance	at 120°C for 5 hours. In saturated atmosphere with trichloroethylene for 1 day. at 50°C, 90% RH for 1 day.	60
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Table 1 (contd.)

	Anionic colloid materials	Anion modifying agents	
Example 4	Maleic acid-modified polyvinyl alcohol	glycine	
5 Example 5	Carrageenan	sulfamic acid	5
Example 6			
10	0.5 parts of m-methoxyphenol was added to 150 parts of 5% aqueous solution (pH 3.0) of acrylic acid-methyl acrylate copolymer to prepare an aqueous system. 100 parts of the same oily solution as in Example 1 was dispersed in the aqueous system to prepare an emulsion containing oily capsule core material particles which has an average particle size of 4.0 microns. On the other hand, 5 parts of monochloroacetic acid was dissolved in 80 parts of 37% aqueous solution of formaldehyde and then 7 parts of 25% ammonia water and 4 parts of sodium hydroxide were added to the solution. After the resultant mixture was thoroughly agitated, 30 parts of urea was added to it and the reaction mixture was refluxed for 30 minutes. The pH of the product was adjusted to 7.0 to prepare an aqueous solution of an anion-modified urea-formaldehyde resin prepolymer.	10	15
15	21 parts of the thus obtained prepolymer solution was added to the above emulsion. The resultant aqueous system has a pH of 3.4. The aqueous system was heated at 60°C for 3 hours for reaction and then cooled with standing to prepare a capsule dispersion. The capsule dispersion was used to prepare a capsule coated paper sheet in the same manner as in Example 1.		20

20	Examples 7—9	20
	3 capsule dispersions were prepared in the same manner as in Example 6 except that anionic colloid materials as shown in Table 2 were used and the pH of each of the dispersions after adding the prepolymer was adjusted to 3.4. The capsule dispersions were used to prepare capsule coated sheets in the same manner as in Example 1.	

25	Anionic Colloid Materials	25
	Example 7 Half-amidation compound of isobutylene-maleic anhydride copolymer	
	Example 8 Carboxymethylcellulose	
	Example 9 Pectic acid	

30	Example 10	30
	5 parts of urea and 0.5 parts of glycine were added to and dissolved in 150 parts of 5% aqueous solution (pH 3.4) of a half-amidation compound of isobutylene-maleic anhydride copolymer to obtain a homogeneous aqueous solution, 100 parts of the same oily solution as in Example 1 was dispersed in the aqueous solution to prepare an emulsion containing oily capsule core material particles which had an average particle size of 4.0 microns. 13 parts of 37% aqueous solution of formaldehyde was added to the emulsion. The mixture was heated to 55°C under a gentle agitation, 5 parts of 10% aqueous solution of resorcin was added to the mixture after maintaining at 55°C for 1 hour and then the mixture was further heated at 55°C for 2 hours and cooled on standing to obtain a capsule dispersion. The resultant capsule dispersion was used to obtain a capsule coated paper sheet.	
35		35
40	The properties of the thus obtained capsule coated paper sheets were examined with the following manners.	40

45	At first, an acceptor coated sheet was prepared by the steps of adding 65 parts of aluminum hydroxide, 20 parts of zinc oxide, 15 parts of a co-melt of zinc, 3,5-di(α -methylbenzyl)salicylate and α -methylstyrene-ethylene copolymer in which the co-melting ratio was 80/20 and an aqueous solution containing 5 parts of polyvinyl alcohol to 300 parts of water, finely dividing the mixture for 24 hours in a ball-mill to obtain a dispersion, adding to the dispersion a latex of carboxylated styrene-butadiene copolymer in which the solid amount was 20 parts to prepare a coating composition, applying the coating composition on one surface of a base paper sheet of 40 g/m ² in an amount of 5 g/m ² on dry basis and drying the coated paper sheet.	45
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50	1. Color formability:	50
	Each capsule coated paper sheet obtained in Examples and Controls was put on the above acceptor coated sheet so that the coated layers were close to each other to provide samples. The samples were pressed with 100 Kg/cm ² for 1 minute to form a color image. The color density of the image on the acceptor coated surface was measured by a spectrophotometer at 600 nm with the use of magnesium oxide as a standard. The measured reflection ratios are shown in Table 3.	
55		55

60	2. Stability of color forming:	
	Each capsule coated sheet obtained in Examples and Controls was treated by standing under the following conditions:	
	(a) heat resistance (b) solvent resistance (c) humidity resistance	at 120°C for 5 hours. in saturated atmosphere with trichloroethylene for 1 day. at 50°C, 90% RH for 1 day.

The color densities of the samples after color images were formed were measured and the results are shown in Table 3 in terms of the color density decreasing ratios with the following ranks.

- 5 A from 0% to less than 5%
 B from 5% to less than 10%
 C from 10% to less than 20%
 D from 20% to less than 50%
 E above 50%

Table 3

	<i>Colour formability</i>	<i>Stability of Color Forming</i>		
		<i>Heat resistance</i>	<i>Solvent resistance</i>	<i>Humidity resistance</i>
Example 1	9	B	A	A
Control 2	35	E	E	E
Example 2	8	A	A	A
Example 3	9	B	A	A
Example 4	7	A	A	A
Example 5	14	B	C	B
Example 6	10	A	A	A
Example 7	11	A	A	B
Example 8	22	C	C	B
Example 9	19	B	CC	B
Example 10	23	B	C	B

As shown in Table 3, every pressure sensitive copying systems prepared with the use of microcapsule obtained by the process of the invention were superior in heat resistance, solvent resistance and humidity resistance. However, no desired capsule dispersion was obtained in Control 1. Also, with the use of the capsule coating composition obtained in Control 2 a pressure sensitive copying system having desired heat resistance, solvent resistance and humidity resistance was not obtained.

15 Claims

1. A method of making microcapsules comprising:
preparing an aqueous dispersion including hydrophobic material particles dispersed therein,
establishing in said dispersion a system for causing polycondensation to form an anion-modified
amino-aldehyde resin; and
20 performing said polycondensation to deposit said anion-modified aminoaldehyde resin on said
hydrophobic material particles to form microcapsules encapsulating said hydrophobic material therein,
and characterized in that said polycondensation is carried out in the presence of at least one anionic
colloid material.

25 2. A method according to Claim 1, in which said anion-modified aminoaldehyde resin is an
amino-aldehyde resin modified with at least one of sulfamic acid, sulfanilic acid and glycine.

30 3. A method according to Claim 1 or 2, in which said hydrophobic material particles are oil
droplets including at least one chromogenic material dissolved therein.

35 4. A method according to any preceding claim, in which said system for causing
polycondensation comprises a water soluble prepolymer of at least one amine and at least one
aldehyde.

40 5. A method according to Claim 4, in which said system further includes an anion-modifying
agent.

45 6. A method according to Claim 4, in which said prepolymer is an anion-modified prepolymer.

50 7. A method according to Claim 5, in which said anion-modifying agent is used in an amount of 1
to 50 mol.% of the amine portion of said prepolymer.

55 8. A method according to Claim 7, in which said anion-modifying agent is used in an amount of 2
to 20 mol.% of the amine portion of said prepolymer.

60 9. A method according to any preceding claim, in which said anionic colloid material is selected
from natural polymers, semisynthetic polymers and synthetic polymers.

65 10. A method according to claim 9, wherein said anionic colloid material comprises at least one
natural polymer selected from gum arabic, carrageenan, sodium alginate, pectic acid, tragacanth gum,
almond gum and agar.

70 11. A method according to claim 9, wherein said anionic colloid material comprises at least one
semi-synthetic polymer selected from carboxymethylcellulose, sulfonated cellulose, sulfonated
methylcellulose, carboxymethylated starch, and phosphonated starch.

75 12. A method according to claim 9, wherein said anionic colloid material comprises at least one

synthetic polymer selected from polymers having at least one anionic monomer unit, partial amidation compounds and partial esterification compounds thereof and carboxylated polyvinyl alcohol.

13. A method according to claim 12, wherein said polymers having at least one anionic monomer unit comprise acrylic acid, methacrylic acid, maleic acid, crotonic acid, itaconic acid and vinylbenzene sulfonic acid. 5

14. A method according to any preceding claim, in which said anionic colloid material is a copolymer of at least one hydrophobic monomer with at least one of acrylic acid, methacrylic acid and maleic acid. 10

15. A method according to any preceding claim, in which said anionic colloid material is used in 10 an amount of at least 0.5% by weight of the amount of said aqueous dispersion. 10

16. A method according to any preceding claim, in which said anionic colloid material is used in an amount of at least 2% by weight of the amount of said aqueous dispersion. 10

17. A method according to Claim 15 or 16, in which the amount of said anionic colloid material used is not larger than 20% by weight of the amount of said aqueous dispersion. 15

18. A method according to any preceding claim, in which said aqueous dispersion including said system for causing polycondensation and said anionic colloid material is prepared in an acidic state. 15

19. A method according to any preceding claim, in which said polycondensation is carried out at a temperature within the range of 20 to 100°C. 15

20. A method according to claim 19, in which said polycondensation is carried out at a 20 temperature within the range of 35 to 70°C. 20

21. A method according to claim 1, substantially as described herein with reference to any of Examples 1 to 10.